

# Quantification of Urea in Ethanol by Raman Spectroscopy

Urea is widely employed as a nitrogen-release fertilizer with more than 90% of urea production destined for agricultural applications[1]. Urea is also known to form complexes with fatty acids[2], which have been employed for separation of complex mixtures and purification processes[3]. In this

application note, we present the quantification of the concentration of urea in ethanol by Raman Spectroscopy and show how this method can be employed for determining the percentage of urea in a solid inclusion compound with stearic acid[4].

Find more information in the video:

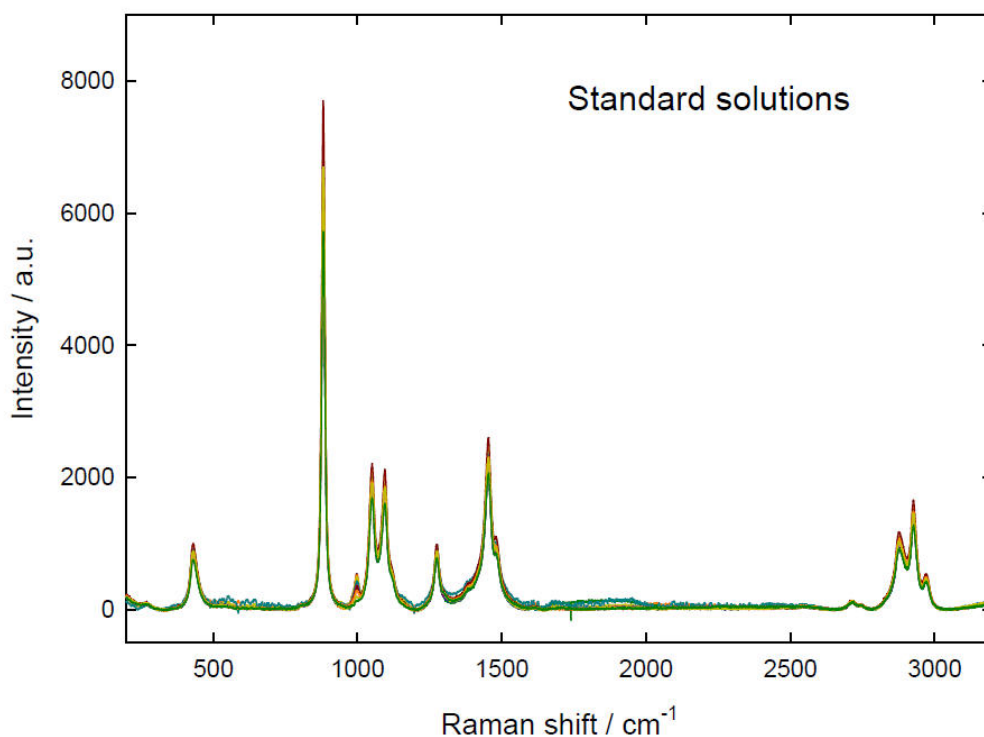
## EXPERIMENTAL

Chemicals: Urea (Aldrich, >99% ); Stearic acid (SA) (Aldrich, >95%); Ethanol (Soria)

Instrument: [i-Raman® Plus 785S](#)

Stock solutions of urea (0.0420 gr/gr ethanol) and SA (0.04126 gr/gr ethanol) were prepared. Standard solutions were prepared by mixing these stock solutions in different proportions in order to obtain urea concentrations between 0 and 0.042 gr/gr ethanol and keeping the total mass concentration

(urea + SA) approximately constant. Raman spectra of 0.5 mL of each solution in aluminum containers were measured employing a 785 nm laser (power: 30%, ~ 90 mW) and 5000 msec of acquisition time (20 repetitions). No effects of heating or evaporation were observed. Spectra of the standard solutions that were background corrected employing the [BWSpec®](#) software are shown in **Figure 1**.



**Figure 1.** Dark-subtracted baseline-corrected Raman spectra of the standard solutions of urea and SA in ethanol.

Spectra were normalized by employing the intensity of the ethanol band at  $1049\text{--}1050\text{ cm}^{-1}$ . Normalized spectra (**Figure 2**) clearly show that the only appreciable change due to the increment in relative urea concentration occurs at the band at  $996\text{--}997\text{ cm}^{-1}$ , which is attributed to urea. This band corresponds to the symmetric C-N stretching[5] that is experimentally and theoretically reported at about  $1010\text{ cm}^{-1}$  for the solid urea,[6] but shifts to lower wavenumbers in solution.[5,7]

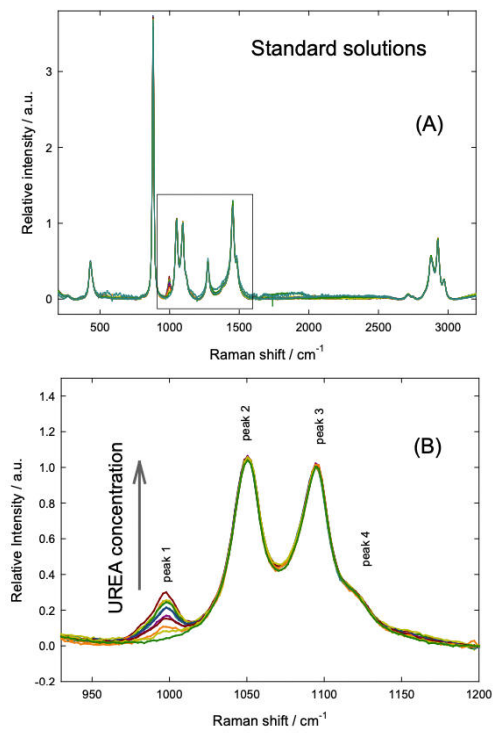
For quantification purposes, spectra were deconvoluted, fitting the experimental results in the region  $950\text{--}1200\text{ cm}^{-1}$  by 4 Lorentzian functions. These curve fitting results are shown in **Figure 3** for some of the standard solutions.

The ratio of the intensities of the fitted peaks assigned to urea at  $996\text{ cm}^{-1}$  (peak 1,  $a_1$ ) and ethanol at  $1049\text{ cm}^{-1}$  (peak 2,  $a_2$ ) was employed as the analytical

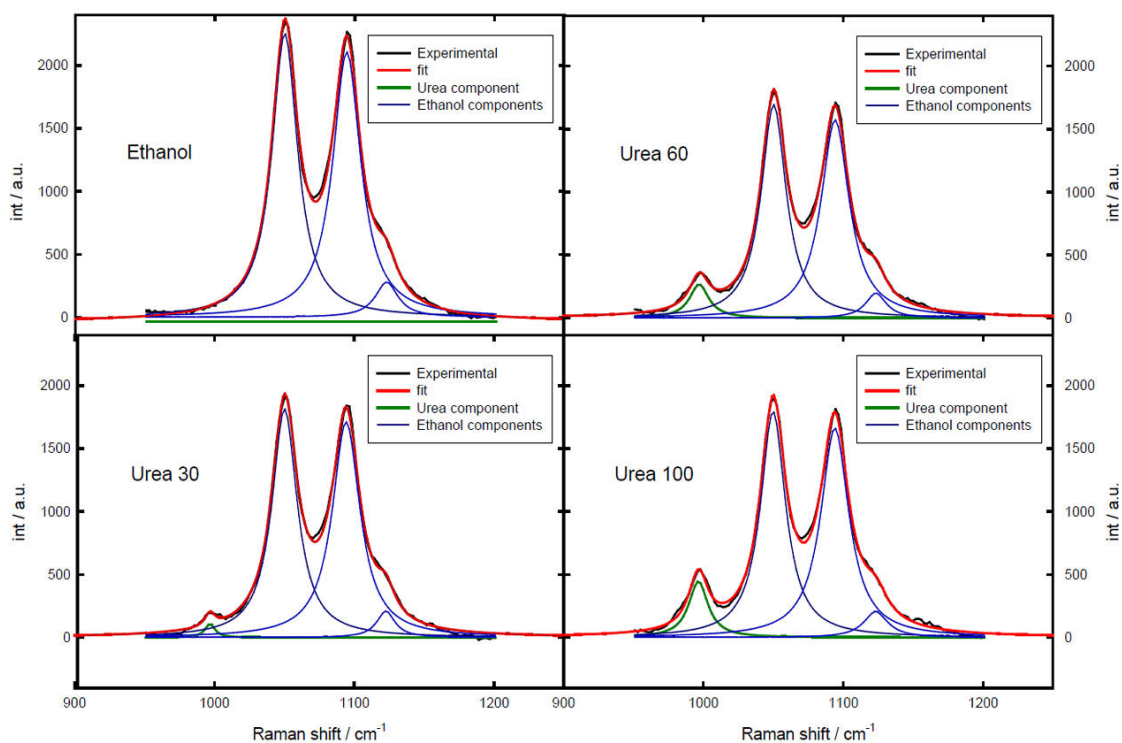
parameters. The dependence of this ratio on the urea concentration of the samples is presented in **Figure 4**. The calibration curve plotted in this figure shows a good linear behavior which indicates this parameter could be employed for urea quantification.

For the determination of the urea content of real samples containing both urea and SA, the solid sample was dissolved in ethanol ( $0.04299\text{ gr/gr}$  ethanol) and Raman spectra were recorded in the same conditions. From the values of the ratio  $a_1/a_2$  for this sample, obtained by fitting of the peaks at  $996$  and  $1049\text{ cm}^{-1}$  (**Figure 5**), the urea concentration of the solution was determined to be  $0.03274\text{ gr urea /gr ethanol}$ . Thus, the urea content of the sample resulted to be  $76\% \text{ w/w}$ . This value is consistent with other reported values for the inclusion compounds formed by stearic acid and urea (ca.  $75\%$ ).[2,4]

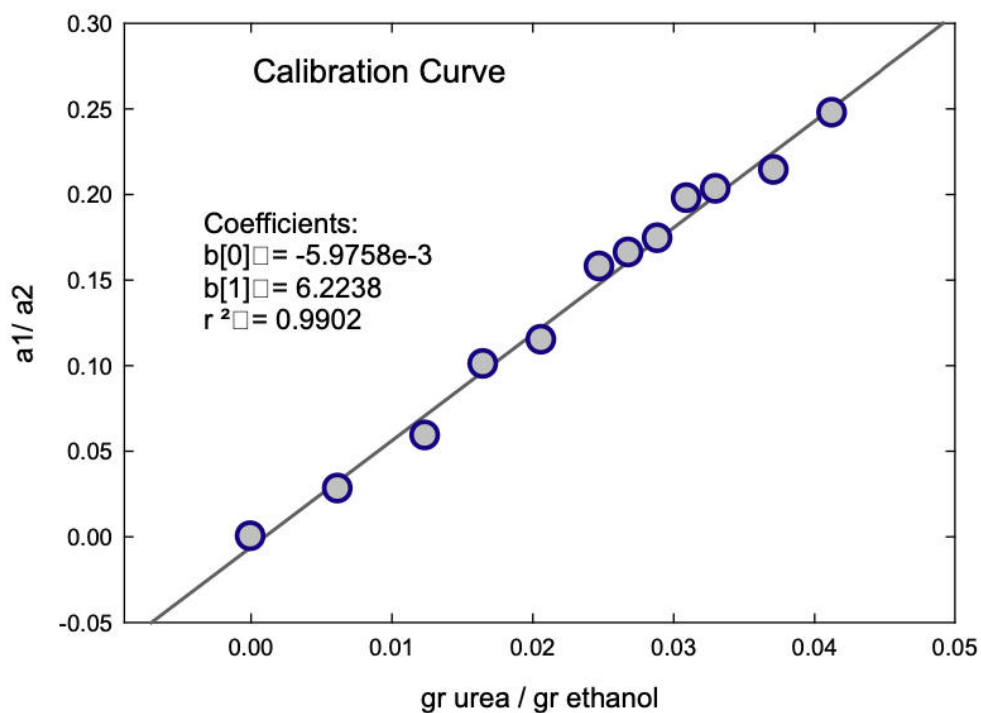
## RESULTS



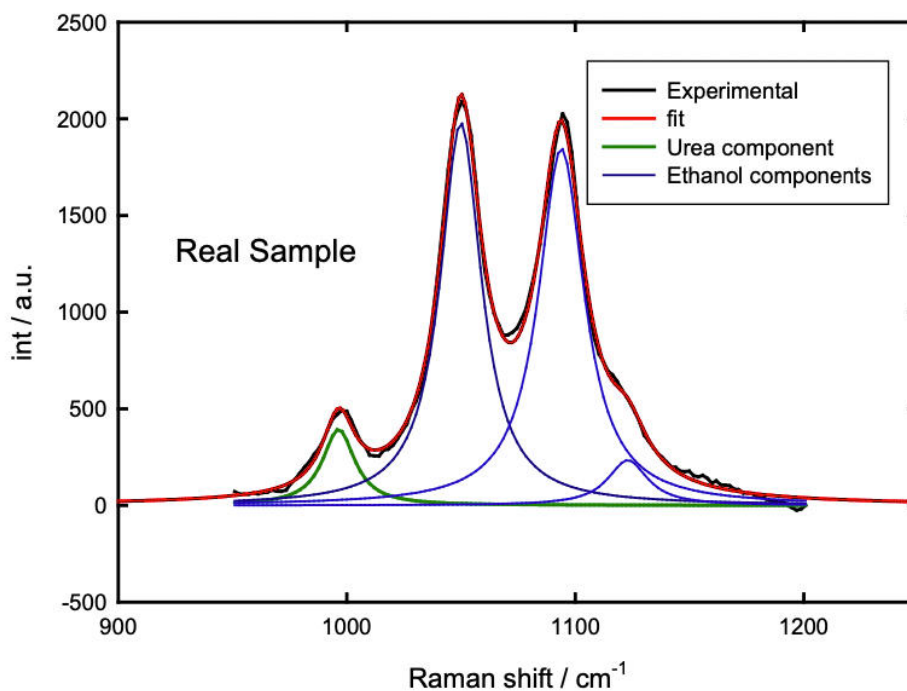
**Fig.2:** Normalized spectra of the standard solutions of urea + SA in ethanol. (A) Complete spectra (B) Analyzed region.



**Figure 3.** Fitting of the Raman spectra in the region 950-1200 cm<sup>-1</sup>. Urea 30, Urea 60 and Urea 100 mean solutions 0.0123, 0.0248 and 0.0413 gr urea/gr ethanol respectively.



**Figure 4.** Calibration curve for the urea quantification in ethanol. Ratio of the intensities of the fitted bands of urea (a1) and ethanol (a2) as a function of the urea content of the standard solutions.



**Figure 5.** Fitting of the sample spectrum.

## CONCLUSION

We presented a simple method for quantification of urea concentration in ethanolic solutions by Raman spectroscopy. The calibration curve presents good linearity in the concentration range analyzed (up to 0.042 gr urea / gr ethanol). The presence of stearic

acid in the samples does not modify appreciably the Raman spectra (at least up to 0.042 gr / gr ethanol), so this method allows the quantification of urea in solid binary samples containing both urea and stearic acid.

## FURTHER INFORMATION

### Related application notes

[Choosing the Most Suitable Laser Wavelength](#)

[Quantification of methanol in contaminated spirits with Raman](#)

## ACKNOWLEDGEMENTS

We would like to thank Dr. Waldemar A. Marmisollé of the Soft Matter Laboratory INIFTA – Universidad

Nacional de La Plata (UNLP), Buenos Aires, Argentina – CONICET for sharing these research results.

## REFERENCES

1. J. H. Meessen, H. Petersen, in Ullmann's Encycl. Ind. Chem., Wiley-VCH Verlag GmbH & Co. KGaA, 2000.
2. H. Schlenk, R. T. Holman, J. Am. Chem. Soc. 1950, 72, 5001–5004.
3. D. G. Hayes, Y. C. Bengtsson, J. M. Van Alstine, F. Setterwall, J. Am. Oil Chem. Soc. 1998, 75, 1403–1409.
4. A. Strocchi, G. Bonaga, Chem. Phys. Lipids 1975, 15, 87–94.
5. R. Keuleers, H. O. Desseyn, B. Rousseau, C. Van Alsenoy, J. Phys. Chem. A 1999, 103, 4621.
6. B. Rousseau, C. Van Alsenoy, R. Keuleers, H. O. Desseyn, J. Phys. Chem. A 1998, 102, 6540–6548.
7. D. Gangopadhyay, S. K. Singh, P. Sharma, H. Mishra, V. K. Unnikrishnan, B. Singh, R. K. Singh, Spectrochim. Acta Part A Mol. Biomol. Spectrosc. 2016, 154, 200–206.

## CONTACT

Metrohm Suisse SA  
Industriestrasse 13  
4800 Zofingen

[info@metrohm.ch](mailto:info@metrohm.ch)

## CONFIGURATION



### Spectromètre Raman portable i-Raman Plus 785S

L'i-Raman® Plus - 785S appartient à notre série primée de spectromètres Raman portables « i-Raman », équipée de notre technologie de spectrométrie intelligente et innovante. Faisant appel à un détecteur à barrette CCD à haute efficacité quantique, avec refroidissement thermoélectrique et une gamme dynamique élevée, ce spectromètre Raman portable fournit des performances exceptionnelles avec un bruit réduit, même avec un temps d'intégration atteignant 30 minutes. Ainsi, les signaux Raman faibles peuvent aussi être mesurés.

Le i-Raman Plus 785S offre la combinaison exclusive d'une large gamme spectrale et d'une haute résolution avec des configurations autorisant des mesures de  $65 \text{ cm}^{-1}$  à  $3350 \text{ cm}^{-1}$ . Le faible encombrement et la légèreté de structure de ce système peu énergivore lui permettent d'effectuer partout des analyses Raman de qualité recherche. Équipé d'une sonde à fibre optique pour faciliter l'échantillonnage, l'i-Raman Plus peut être utilisé avec un support de cuvette, un microscope vidéo, une table de translation XYZ avec support de sonde ainsi qu'avec notre logiciel d'analyse à variantes multiples BWIQ® et le logiciel d'identification BWID®. Avec le i-Raman Plus, vous avez une solution Raman pérenne de haute fidélité pour l'analyse qualitative et quantitative.



### Porte-cuve Raman pour sonde de mesure de 9,5 mm

Le porte-cuve Raman BCR100A vous permet de mesurer facilement le spectre Raman de liquides et de poudres en fixant une sonde Raman au support. Cet accessoire utilise un miroir intérieur doté d'un obturateur de précision à trois points pour une reproductibilité inégalée et amplifie le signal Raman jusqu'à trois fois plus que les porte-cuves classiques. Il est conçu pour que la tige de la sonde n'entre pas directement en contact avec la cuve et comporte un piège à lumière pour réduire la fluorescence de fond. Le BCR100A est disponible en modèles pour sondes de 9,5 mm ou 12 mm de diamètre et peut être utilisé pour le prélèvement dans des liquides ou des poudres avec tout type de cuve standard de 12,5 mm x 12,5 mm de diamètre extérieur (course de 1 cm).