



Application Note AN-PAN-1032

Monitoring corrosion in power plants with online process analysis

Faster ultratrace measurements of iron (Fe) and copper (Cu)

Corrosion in the water-steam circuit of power plants results in shorter lifetimes of most metal components and can lead to potentially dangerous situations. Flow Accelerated Corrosion (FAC) leads to thinned pipes and elevated iron concentrations in the circuit. Additionally, metal transport issues such as with copper from «copper heat exchangers» can lead to deposition on the high-pressure turbine blades. Current methods can monitor these issues but cannot prevent them as analysis times are extremely long (up to three weeks).

This Process Application Note details the online ultratrace analysis of iron and copper in power plants. This method offers results in 20 minutes, meaning faster response times for out of specification readings. In combination with the power plant's Distributed Control System (DCS), online monitoring of these analytes using a process analyzer ensures that corrosion can be controlled before it affects the power plant efficiency, ultimately decreasing downtime and lowering maintenance costs.

In power plants, corrosion is the primary factor leading to costly and critical downtimes. The water-steam circuits in fossil and nuclear power plants are inherently prone to corrosion, as metal components are constantly in contact with water. Corrosion leads to shorter lifetimes for the carbon steel pipework and copper (Cu) heat exchangers, among other issues. At high temperatures, steam reacts with the iron (Fe) in the carbon steel of steam boilers and forms a thin

The diagram illustrates the corrosion process on a metal surface, showing the interaction between the metal, an oxide film, and the surrounding fluid. The metal is labeled "CORRODING METAL" at the bottom. The fluid above is labeled "Fluid Flow" with an arrow indicating movement to the right. The diagram is divided into three horizontal sections: "Diffusion" at the top, "Oxide Film Formation" in the middle, and "Corrosion" at the bottom. The "Diffusion" section shows the movement of species (represented by blue wavy arrows and red dots) through the fluid. The "Oxide Film Formation" section shows the growth of an oxide film (represented by red dots) on the metal surface. The "Corrosion" section shows the dissolution of the metal (represented by red dots) and the release of "Particulate Oxide Release" (red dots) into the fluid. The "Corrosion" section also shows "Oxide Dissolution" (blue wavy arrows) and "Hydrogen Production" (white arrow pointing up). The diagram is numbered 1 through 6, with lines connecting the numbers to the corresponding processes.

Fluid Flow →

Diffusion

Oxide Film Formation

Corrosion

Particulate Oxide Release

Oxide Dissolution

Hydrogen Production

CORRODING METAL

The underlying metal corrodes to re-create the oxide and thus Fe loss continues, potentially leading to catastrophic failure in the piping. In power plants which utilize Cu alloy heat exchanger tubes in the condensate system, Cu corrosion and transportation is also an issue, leading to Cu deposition on high pressure turbine blades and loss of performance. Corrosion and metal transport increase with power output over a certain threshold, and therefore so does the deposition onto the turbines. Considering up to 10% loss of efficiency from the turbine blades, the power output will still be the same, but 10% more energy has to be consumed, and as flow rate increases, corrosion also increases.

Determining optimal power output with minimal FAC is important not only for saving costs but also for the

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The metal transportation in power plant water circuits is currently monitored by CPS racks which collect particulate metals on filter pads over a period of one day to a week. The pads are later digested, and the metals analyzed by ICP-OES or ICP-MS. Total analysis time can take from one to three weeks. Monitoring only the accumulated corrosion products causes a loss of transportation peaks, and detailed information on why the metal loss occurred is lost. A maximum of 2 µg/L Fe is recommended by the Electric Power Research Institute (EPRI) in order to avoid FAC-related issues in the water-steam circuit, and these levels are not accurately measured with the current CPS racks, as seen in long-term comparisons. The continuous online ultratrace analysis of Fe and Cu in the water-steam circuit of power plants is possible using the **2060 Process Analyzer** (Figure 2) from Metrohm Process Analytics. This automated process analysis system enables early detection of corrosion processes and peaks, and also monitors the formation and destruction of the protective oxide layer (Figure 3). Continuous analyses also signal a problem before dissolved metals can reach the condensate stream and thus the turbine blades where they would cause damage. In combination with the power plant's Distributed Control System (DCS), online monitoring of Fe and Cu ensures that corrosion can be controlled before it affects the power plant efficiency, ultimately decreasing downtime and lowering maintenance costs.



Figure 2. 2060 Process Analyzer for photometric measurements of ultratrace iron and copper in the water-steam circuit of power plants.

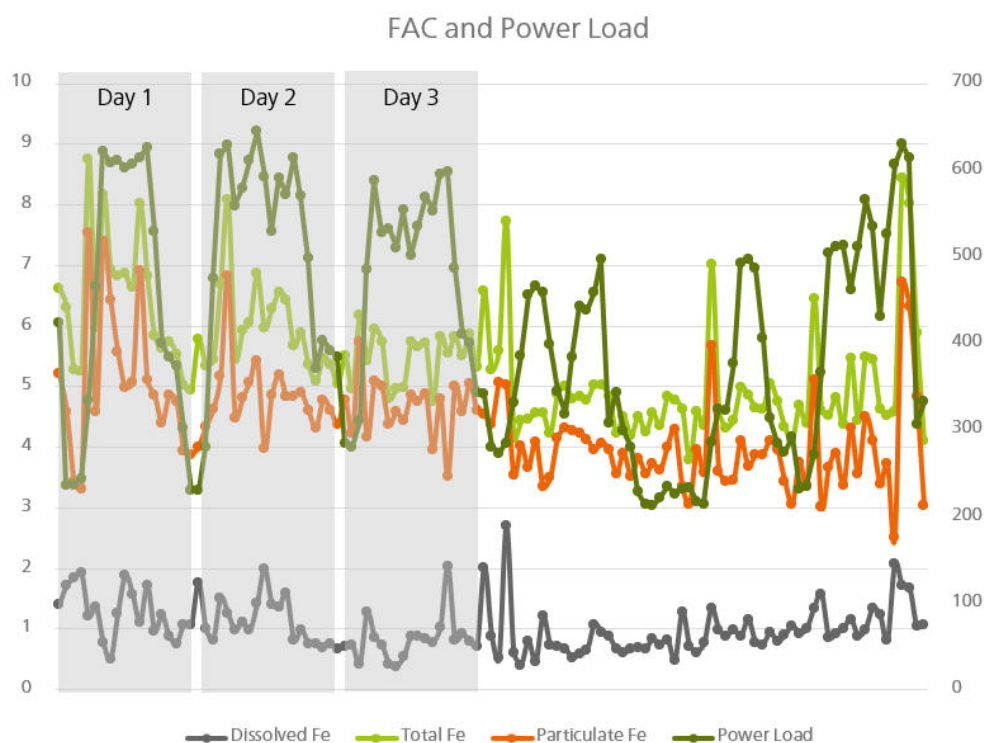


Figure 3. Data obtained from a Metrohm Process Analyzer in a power plant, used to monitor dissolved, total, and particulate iron (in µg/L), against power load (MW).

APPLICATION

Iron and copper content are determined by sample acid digestion followed by photometric determination using TPTZ and Bicinchoninate as color reagents respectively using the **2060 Process Analyzer** from Metrohm Process Analytics (**Figure 2**). Metal

complexes such as magnetite, hematite, iron oxide (Fe_2O_3), and iron hydroxide ($\text{Fe}(\text{OH})_2$) are dissociated into their dissolved forms by using nitric acid (HNO_3). The total analysis time is 20 minutes.

Table 1. Parameters to monitor in the water-steam circuit of power plants.

Analyte	Concentration (µg/L)
Fe(II, III) and $\text{Fe}(\text{OH})_2$	0–10
Total Fe	0–10
Cu	0–11

REMARKS

This analyzer is built based on the International Association for the Properties of Water and Steam Technical Guidance Document: Corrosion Product Sampling and Analysis for Fossil and Combined Cycle Plants [2]. The ranges listed above (**Table 1**) are typically very low and may not reflect the expected values due to the fact that CPS racks cannot measure with the same accuracy.

It is expected that many power plants currently have

much higher levels of dissolved Fe and Cu in the water-steam circuit, causing problems. The measurement ranges for the dissolved metals can be easily expanded for this reason.

Other online applications are available for power plants such as: calcium and sulfate in flue gas scrubbing, boric acid in the Primary Water Circuit, amine concentration and CO₂ loading, silica in boiler feed water, and more.

FURTHER READING

Related application documents

[WP-076: Process analyzers as proactive solutions for online corrosion monitoring](#)

[AN-PAN-1045: Online monitoring of copper corrosion inhibitors in cooling water](#)

BENEFITS FOR PHOTOMETRY IN PROCESS

- Protection of company assets with built in alarms at specified warning limits to prevent corrosion
- Safer working environment for employees (corrosive environments)
- Guarantee compliance with environmental standards



REFERENCES

1. Dooley, B.; Lister, D. Flow-Accelerated Corrosion in Steam Generating Plants. **2018**, 51.
2. IAPWS Technical Guidance Document: Corrosion Product Sampling and Analysis for Fossil and Combined Cycle Plants <http://www.iapws.org/techguide/CorrosionSampling.html> (accessed 2021 -12 -16)

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